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TETRAHEDRON

Senepodines B–E, new C₂₂N₂ alkaloids from Lycopodium chinense

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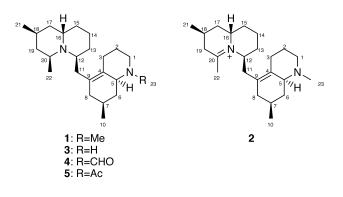
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Abstract—Four new $C_{22}N_2$ *Lycopodium* alkaloids, senepodines B–E (**2–5**), consisting of an octahydroquinoline ring and a quinolizidine ring have been isolated together with senepodine A (**1**) from the club moss *Lycopodium chinense*. The relative and absolute stereochemistry of **2-5** were determined by combination of NOESY correlations and chemical transformation, while the absolute configuration of **1** was assigned by exciton chirality method. © 2003 Elsevier Science Ltd. All rights reserved.

A number of Lycopodium alkaloids with fascinating heterocyclic frameworks of C16N, C16N2, and C27N3 types have been isolated from club moss of the genus Lycopodium.¹ These unique skeletons have attracted great interest not only from biogenetic^{1,2} and biological³ points of view but also as challenging targets for total synthesis.⁴ Recently, we have isolated serratezomine A⁵ with a secoserratinine-type skeleton from Lycopodium serratum var. serratum, complanadine A^6 with a lycodine-dimeric skeleton, lyconadin A⁷ from L. complanatum, and lyconesidine A^8 and senepodine A^9 from Lycopodium chinense. Biomimetic transformation from serratinine into serratezomine A through a modified Polonovski reaction has also been reported.¹⁰ Our interest has been focused on isolation of structurally interesting alkaloids and biosynthetic intermediates to clarify the biogenetic pathway. Further investigation on extracts of L. chinense (Lycopodiaceae) resulted in the isolation of four new C₂₂N₂ alkaloids, senepodines B–E (2–5), as well as a known related alkaloid, senepodine A (1).⁹ This paper describes the isolation and structure elucidation of 2-5 and the absolute configuration of 1.

Isolation of senepodines B (2)–E (5). The club moss of L. chinense was extracted with MeOH, and the extract was partitioned between EtOAc and 3% tartaric acid. Watersoluble materials, which were adjusted at pH 10 with sat. Na₂CO₃, were extracted with CHCl₃. CHCl₃-soluble materials were subjected to an amino silica gel column (hexane/EtOAc, $1:0\rightarrow0:1$, and then CHCl₃/MeOH, $1:0\rightarrow0:1$), in which a fraction eluted with hexane/EtOAc (3:2) was purified by a silica gel column (CHCl₃/MeOH \rightarrow CHCl₃/MeOH/TFA) and then C₁₈ HPLC to afford senepodines B (2, 0.0004%), C (3, 0.003%), D (4, 0.1%), and E (5,

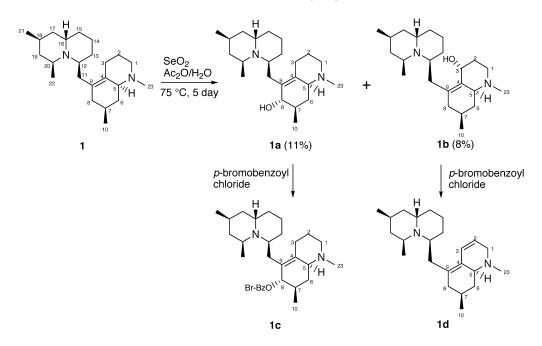
0.01%) together with senepodine A (1, 0.003%) and known C₁₆N type alkaloids, lyconesidines A⁸ (0.002%), B⁸ (0.005%), and C⁸ (0.003%), lycodoline¹¹ (0.001%), and lucidine B¹² (0.001%).



Absolute stereochemistry of senepodine A (1). Although the gross structure and the relative stereochemistry of senepodine A (1) have been reported previously,¹⁰ the absolute stereochemistry remains to be defined. The absolute stereochemistry of 1 was elucidated by applying exciton chirality method¹³ after oxidation of allylic position at C-3 or C-8 with selenium dioxide, followed by introduction of p-bromobenzoyl group into the hydroxyl group at C-3 or C-8 as follows. Treatment of 1 with selenium dioxide in acetic anhydride and H₂O at 75°C for 5 days gave two oxidized products (1a, 11% yield; 1b, 8%) (Scheme 1). The oxidative position and its relative stereochemistry of the hydroxy group of 1a and 1b were elucidated by NOESY correlations (Fig. 1) to be 8α and 3α , respectively. Inspection of molecular models giving the same results was obtained by Monte Carlo simulation¹⁴ using MMFF force field.¹⁵ Treatment of **1a** with *p*-bromobenzoyl chloride

Keywords: Lycopodium chinense; alkaloids; exciton chirality method.

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Scheme 1.

afforded the corresponding *p*-bromobenzoyl ester (1c), whereas in the case of 1b with the axially oriented hydroxyl at C-3, dehydration proceeded by the same treatment to give

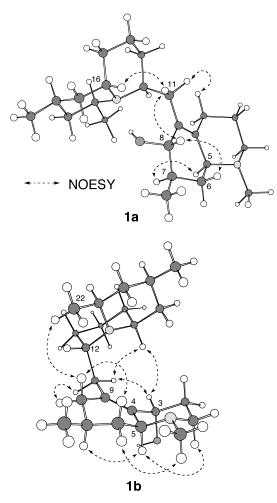


Figure 1. Selected NOESY correlations for 8-hydroxysenepodine A (1a) and 3-hydroxysenepodine A (1b).

the diene (1d). Sign of the first Cotton effect [λ_{max} 244 nm ($\Delta \varepsilon -9.2$)] for 1c was negative (Fig. 2), indicating that the chirality between the *p*-bromobenzoyl group at C-8 and the olefin at C-4-C-9 was as shown in Figure 2 (left-handed screw). Thus, the absolute configuration at C-8 of the oxidative compound (1a) was assigned as *S*, indicating that the absolute structure of senepodine A (1) was as shown in Scheme 1.

Structures of senepodines B-E (2–5). Senepodine B (2) was shown to have the molecular formula of $C_{23}H_{39}N_2$ by HRFABMS [*m*/*z* 343.3133, (M)⁺, Δ +2.0 mmu]. The IR spectrum was indicative of the presence of imine (1680 cm⁻¹) functionality. ¹H and ¹³C NMR data including DEPT experiments (Tables 1 and 2) disclosed three sp^2

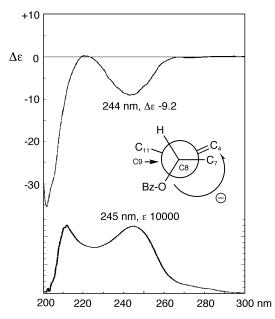


Figure 2. CD and UV spectra of *p*-bromobenzoate (1c) of 8-hydroxysenepodine A (1a) and rotation model for the C8–C9 bond.

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quaternary carbons, five sp^3 methines, eleven sp^3 methylenes, and four methyl groups, and the chemical shift [δ_C 189.17 (s)] of C-20 remarkably shifted to lower field as compared with that [δ_C 50.14 (d)] of **1**. The presence of an iminium carbon (C-20) was elucidated by HMBC correlations for H₃-22 to C-20. In addition, the ¹H signals at 12-, 16-, 19-, and 22-positions and the ¹³C signals at 12-, 16-, and 22-positions around the imine functionality were observed at lower field due to deshielding effects (Tables 1 and 2). 2D NMR data of **2** including the ¹H–¹H COSY, HOHAHA, HMQC, and HMBC spectra, and FABMS/MS fragmentation (Fig. 3) corroborated well with those of the imine (C-20 and N) form of **1**. Treatment of **2** with NaBH₄ afforded senepodine A (**1**) by stereoselective attack of a hydride outside the cage form of **1**.¹⁰

HRFABMS data [*m*/*z* 331.3108, (M+H)⁺, Δ -0.5 mmu] of senepodine C (**3**) established the molecular formula to be C₂₂H₃₈N₂, which was smaller than that of senepodine A (**1**) by a CH₂ unit. ¹H and ¹³C NMR data (Tables 1 and 2, respectively) of **3** were analogous to those of **1** in the quinolizidine part, although a methyl signal ($\delta_{\rm H}$ 2.30) at C-23 lacking for **3** was observed for **1**. The gross structure of

Table 1. ¹H NMR data [δ_H (*J*, Hz)] of senepodines A (1), B (2), and C (3) in CD₃OD at 300 K

1a 2.31 (1H, m) 3.19 (1H, m) 3.09 (1H, ddd, 12.9, 12.8, 3.0) 1b 2.92 (1H, brd, 11.8) 3.53 (1H, m) 3.38 (1H, brd, 12.8) 2a 1.54 (1H, m) 1.74 (1H, m) 1.66 (1H, m) 2b 1.73 (1H, m) 2.02 (1H, m) 2.00 (1H, m) 3a 1.75 (1H, m) 2.02 (1H, m) 2.93 (1H, brd, 14.5) 3b 2.86 (1H, brd, 13.6) 2.98 (1H, m) 2.05 (1H, m) 4		1	2	3
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1a	2.31 (1H, m)	3.19 (1H, m)	
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1b	2.92 (1H, brd, 11.8)	3.53 (1H, m)	3.38 (1H, brd, 12.8)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	2a	1.54 (1H, m)	1.74 (1H, m)	1.66 (1H, m)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	2b	1.73 (1H, m)	2.02 (1H, m)	2.00 (1H, m)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	3a	1.75 (1H, m)	2.02 (1H, m)	2.93 (1H, brd, 14.5)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		2.86 (1H, brd, 13.6)	2.98 (1H, m)	2.05 (1H, m)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	5		3.82 (1H, m)	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	6a		1.35 (1H, m)	
	6b	2.09 (1H, m)	2.34 (1H, m)	2.11 (1H, m)
	7	1.58 (1H, brd, 14.0)	1.69 (1H, m)	1.75 (1H, brd, 13.8)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	8a	1.79 (1H, m)	1.96 (1H, m)	
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrr$	8b	1.92 (1H, brd, 16.1)	2.09 (1H, brd, 15.6)	2.02 (1H, brd, 16.8)
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	9			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	10	0.99 (3H, d, 6.6)	1.07 (3H, d, 6.4)	1.03 (3H, d, 6.4)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	11a	2.13 (1H, brd, 12.5)	2.43 (1H, brd, 11.5)	2.37 (1H, brd, 12.2)
	11b	3.03 (1H, dd, 12.5,	3.11 (1H, dd, 11.5,	3.17 (1H, dd, 12.5,
$ \begin{array}{llllllllllllllllllllllllllllllllllll$		12.2)	11.1)	12.2)
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrr$			4.70 (1H, m)	3.79 (1H, m)
$ \begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrr$	13a	1.34 (1H, brd, 12.6)		1.55 (1H, brd, 10.6)
$ \begin{array}{lllllllllllllllllllllllll$	13b	1.70 (1H, m)	1.87 (1H, m)	1.84 (1H, m)
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	14a	1.68 (1H, m)	1.72 (1H, m)	1.70 (1H, m)
$ \begin{array}{lllllllllllllllllllllllllllllll$	14b	1.77 (1H, m)	1.87 (1H, m)	1.83 (1H, m)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		1.28 (1H, brd, 15.5)	1.88 (1H, m)	1.60 (1H, brd, 16.1)
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	15b		1.88 (1H, m)	2.17 (1H, m)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	16	3.46 (1H, m)	4.28 (1H, m)	3.85 (1H, brd, 13.3)
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	17a		1.73 (1H, m)	1.60 (1H, m)
19a 1.13 (1H, ddd, 12.5, 2.54 (1H, m) 1.38 (1H, m) 12.5, 12.5) 10b 1.72 (1H, m) 3.03 (1H, m) 19b 1.72 (1H, m) 3.03 (1H, m) 1.93 (1H, m) 20 3.44 (1H, m) 3.89 (1H, m) 3.89 (1H, m) 21 0.90 (3H, d, 6.5) 1.04 (3H, d, 6.3) 0.94 (3H, d, 6.3) 22 1.18 (3H, d, 6.1) 2.49 (3H, s) 1.38 (3H, d, 6.3)	17b	1.59 (1H, brd, 12.8)	1.87 (1H, m)	1.70 (1H, m)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	18	1.82 (1H, m)	1.96 (1H, m)	1.98 (1H, m)
19b 1.72 (1H, m) 3.03 (1H, m) 1.93 (1H, m) 20 3.44 (1H, m) 3.89 (1H, m) 21 0.90 (3H, d, 6.5) 1.04 (3H, d, 6.3) 0.94 (3H, d, 6.3) 22 1.18 (3H, d, 6.1) 2.49 (3H, s) 1.38 (3H, d, 6.3)	19a	1.13 (1H, ddd, 12.5,	2.54 (1H, m)	1.38 (1H, m)
20 3.44 (1H, m) 3.89 (1H, m) 21 0.90 (3H, d, 6.5) 1.04 (3H, d, 6.3) 0.94 (3H, d, 6.3) 22 1.18 (3H, d, 6.1) 2.49 (3H, s) 1.38 (3H, d, 6.3)		12.5, 12.5)		
21 0.90 (3H, d, 6.5) 1.04 (3H, d, 6.3) 0.94 (3H, d, 6.3) 22 1.18 (3H, d, 6.1) 2.49 (3H, s) 1.38 (3H, d, 6.3)	19b	1.72 (1H, m)	3.03 (1H, m)	1.93 (1H, m)
22 1.18 (3H, d, 6.1) 2.49 (3H, s) 1.38 (3H, d, 6.3)	20	3.44 (1H, m)		3.89 (1H, m)
	21	0.90 (3H, d, 6.5)	1.04 (3H, d, 6.3)	
23 2.30 (3H, s) 2.89 (3H, s)	22	1.18 (3H, d, 6.1)	2.49 (3H, s)	1.38 (3H, d, 6.3)
	23	2.30 (3H, s)	2.89 (3H, s)	

Table 2. ^{13}C NMR data (δ_C) of senepodines A (1), B (2), and C (3) in CD_3OD at 300 K					
1	2	3			

	1	2	3
1	58.22	57.00	45.36
2	26.53	24.48	24.48
3	29.17	27.19	27.01
4	131.74	128.62	128.82
5	66.02	66.59	57.61
6	38.95	35.52	37.01
7	29.02	28.93	28.49
8	40.45	39.17	39.67
9	130.91	135.24	133.66
10	22.44	21.66	21.58
11	35.37	35.13	33.21
12	53.88	62.07	55.89
13	19.24	28.58	19.51
14	20.07	19.66	18.41
15	24.54	34.06	24.23
16	52.93	58.45	55.33
17	39.88	35.13	38.13
18	26.10	21.20	25.03
19	44.11	43.80	42.00
20	50.14	189.17	53.32
21	22.44	20.18	21.61
22	20.28	24.54	18.02
23	43.29	41.39	

3 was elucidated by 2D NMR (${}^{1}H{-}{}^{1}H$ COSY, HOHAHA, HMQC, and HMBC) data. Treatment of **3** with CH₃I gave senepodine A (**1**). Thus, the structure of senepodine C was assigned as **3**.

HRFABMS data $[m/z 359.3067, (M+H)^+, \Delta +0.5 \text{ mmu}]$ of senepodine D (4) established the molecular formula, $C_{23}H_{38}N_2O$. The IR spectrum was indicative of the presence of amide carbonyl (1667 cm⁻¹). ¹H and ¹³C NMR data of 4 were analogous to those of senepodine A (1), although ¹H NMR signals were broad. The presence of an amide carbonyl carbon [C-23, δ_C 164.21 (s)] was elucidated by HMBC correlations for H-23 (δ_H 8.07) to C-1 and C-5, and H-1 and H-5 to C-23 through a nitrogen atom. Treatment of 4 with LiAlH₄ afforded senepodine A (1), while hydrolysis of 4 with 1N HCl at 100°C for 6 h gave senepodine C (3) (Scheme 2). Thus, senepodine D (4) was concluded to be the *N*-formyl form at C-23 of senepodine A (1).

HRFABMS data [*m*/*z* 373.3223, (M+H)⁺, Δ +0.4 mmu] of senepodine E (**5**) established the molecular formula, C₂₄H₄₀N₂O. The IR and NMR spectra were indicative of the presence of an *N*-acetyl group (1640 cm⁻¹; $\delta_{\rm H}$ 2.09; $\delta_{\rm C}$ 171.88). Acetylation of senepodine C (**3**) afforded senepodine E (**5**) (Scheme 2). Thus, senepodine E (**5**) was concluded to be the *N*-acetyl form at C-23 of senepodine A (**1**).

Plausible biogenesis of senepodines B-E (2–5). Senepodines B-E (2–5) are $C_{22}N_2$ Lycopodium alkaloids, consisting of an octahydroquinoline and a quinolizidine ring. A plausible biogenetic pathway for senepodines B (2)–E (5) as well as senepodine A^{10} (1) is proposed as shown in Scheme 3. Biogenetically, the decahydroquinoline and quinolizidine units in 2–5 may be derived from an intermediate A with loss of a carbon (path **a**), which have been proposed previously,¹⁰ although an alternative path **b**,

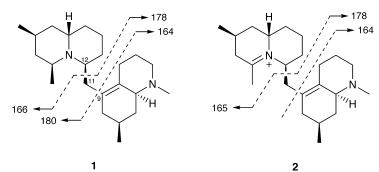
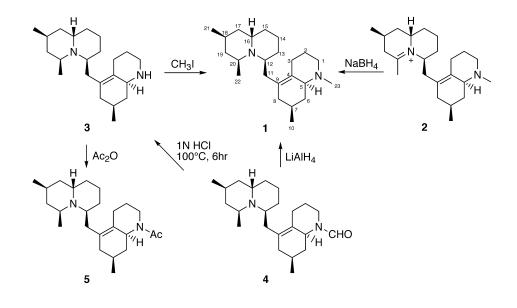
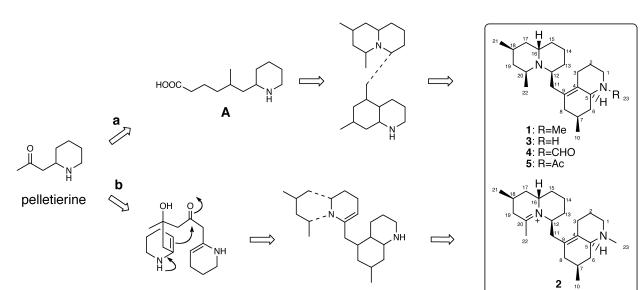


Figure 3. Fragmentation patterns observed in positive ion FABMS/MS spectra of senepodines A (1) (precursor ion, m/z 344) and B (2) (m/z 343).



Scheme 2.



phlegmarane skeleton

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which form the C-20–N and C-17–C-16 connectivities of senepodines B–E with a C6 unit, is also possible.

Bioactivity of senepodines A-E (1–5). Senepodines A-E (1–5) exhibited cytotoxicity against murine lymphoma L1210 cells (IC₅₀ 0.1, 0.1, 1.2, 0.8, and 0.6 µg/ml, respectively) in vitro.

1. Experimental

1.1. General methods

¹H and 2D NMR spectra were recorded on a 600 MHz spectrometer at 300 K, while ¹³C NMR spectra were measured on a 150 MHz spectrometer. Each NMR sample of senepodines A-E (1-5) was prepared by dissolving 1.0 mg in 30 µl of CD₃OD in 2.5 mm micro cells (Shigemi Co. Ltd.) and chemical shifts were reported using residual CD₃OD (δ_H 3.31 and δ_C 49.0) as an internal standard. Standard pulse sequences were employed for the 2D NMR experiments. 1H-1H COSY, HOHAHA, and NOESY spectra were measured with spectral widths of both dimensions of 4800 Hz, and 32 scans with two dummy scans were accumulated into 1K data points for each of 256 t_1 increments. NOESY and HOHAHA spectra in the phase sensitive mode were measured with a mixing time of 800 and 30 ms, respectively. For HMQC spectra in the phase sensitive mode and HMBC spectra, a total of 256 increments of 1 K data points were collected. For HMBC spectra with Z-axis PFG, a 50 ms delay time was used for long-range C–H coupling. Zero-filling to 1 K for F_1 and multiplication with squared cosine-bell windows shifted in both dimensions were performed prior to 2D Fourier transformation. FABMS was measured by using glycerol as a matrix.

1.2. Material

The club moss *L. chinense* was collected at Kiyosato in Hokkaido in 2001. The botanical identification was made by Mr N. Yoshida, Health Sciences University of Hokkaido. A voucher specimen has been deposited in the herbarium of Hokkaido University.

1.3. Extraction and isolation

The club moss (2 kg) of *L. chinense* was extracted with MeOH (10 l×3). The MeOH extract (146 g) was partitioned between EtOAc and 3% tartaric acid. Water-soluble materials, after being adjusted at pH 10 with sat. Na₂CO₃, were partitioned with CHCl₃. CHCl₃-soluble materials (2.4 g) were subjected to an amino silica gel column (hexane/EtOAc, 1:0 \rightarrow 0:1, and then CHCl₃/MeOH, 1:0 \rightarrow 0:1). The fraction eluted with hexane/EtOAc (3:2) was separated by a silica gel column (CHCl₃/MeOH/EtOAc, 10:1:0.5 \rightarrow MeOH/TFA, 10:1) and, then C₁₈ HPLC (Mightysil RP-18 GP 250-20, 5 µm, Kanto Chemical Co., 10×250 mm; eluent, 14–35% CH₃CN/0.1–0.2% TFA; flow rate, 2 ml/min; UV detection at 205 nm) to afford senepodine A (1, 0.003%), B (2, 0.0004%), C (3, 0.003%), D (4, 0.1%), and E (5, 0.01%), together with

lyconesidines A (0.002%), B (0.005%), C (0.003%), lycodoline (0.001%), and lucidine B (0.001%).

1.3.1. Senepodine B (2). Colorless solid; $[\alpha]_{D}^{27} = -38^{\circ}$ (*c* 0.2, MeOH) (with TFA); IR (neat) ν_{max} 3450, 3295, 2927, 1680, 1195, and 1128 cm⁻¹; ¹H and ¹³C NMR data (Table 1); FABMS *m/z* 343 (M)⁺; HRFABMS *m/z* 343.3133 (M⁺; calcd for C₂₃H₃₉N₂, 343.3113).

1.3.2. Senepodine C (3). Colorless solid; $[\alpha]_D^{25} = -15^\circ$ (*c* 1.0, MeOH); IR (neat) ν_{max} 3260, 2920, 1700, 1450, and 1115 cm⁻¹; ¹H and ¹³C NMR data (Table 1); FABMS *m*/*z* 331 (M+H)⁺; HRFABMS *m*/*z* 331.3108 (M+H; calcd for C₂₂H₃₉N₂, 331.3113).

1.3.3. Senepodine D (4). Colorless solid; $[\alpha]_D^{-3} = -33^\circ$ (*c* 0.5, MeOH); IR (neat) ν_{max} 2920, 1667, 1450, and 1370 cm⁻¹; ¹H NMR (CD₃OD) δ : 0.96 (3H, d, *J*=6.3 Hz), 1.02 (3H, d, *J*=6.1 Hz), 1.20 (1H, m), 1.36 (1H, m), 1.39 (3H, d, *J*=6.2 Hz), 2.70 (1H, dd, *J*=13.0, 8.4 Hz), 3.51 (1H, dd, *J*=12.6, 8.4 Hz), 3.90 (3H, m), 4.62 (1H, m), 8.07 (1H, s); ¹³C NMR (CD₃OD) δ : 18.90, 19.81, 21.97, 22.20, 23.50, 25.47, 26.51, 28.95, 33.96, 35.39, 38.84, 39.91, 40.85, 42.83, 51.99, 54.32, 55.36, 56.23, 132.06, 133.15, 164.21. FABMS *m*/*z* 359 (M+H)⁺; HRFABMS *m*/*z* 359.3067 (M+H; calcd for C₂₃H₃₉N₂O, 359.3062).

1.3.4. Senepodine E (5). Colorless solid; $[\alpha]_{D}^{22} = -106^{\circ}$ (*c* 1.0, MeOH); IR (neat) ν_{max} 2920, 1640, 1430, and 1370 cm⁻¹; ¹H NMR (CD₃OD) δ : 0.87 (3H, d, *J*=6.3 Hz), 0.98 (3H, d, *J*=3.5 Hz), 1.07 (3H, d, *J*=6.0 Hz), 1.34 (2H, m), 1.52 (1H, brd, *J*=12.8 Hz), 2.09 (3H, s), 2.71 (1H, dd, *J*=13.4, 9.0 Hz), 2.89 (1H, dd, *J*=12.0, 12.0 Hz), 2.99 (1H, m), 3.18 (1H, m), 4.70 (1H, m); ¹³C NMR (CD₃OD) δ : 19.59, 21.02, 21.56, 22.41, 22.71, 23.64, 24.60, 25.79, 26.59, 29.01, 35.37, 35.60, 40.30, 40.79, 41.54, 45.26, 51.92, 53.15, 58.04, 130.65, 133.10, 171.88. FABMS *m/z* 373 (M+H)⁺; HRFABMS *m/z* 373.3223 (M+H; calcd for C₂₄H₄₁N₂O, 373.3219).

1.3.5. Oxidation of Senepodine A (1). To a solution of 1 (15.9 mg) in water (1.0 ml) at 75°C was added selenium oxide (1.0 g) and acetic anhydride (75 μ l). The mixture was allowed to stand at 75°C for 5 days. The residue was dissolved in Na₂CO₃ aq. and extracted with CHCl₃. After evaporation of solvent, the residue was applied to C₁₈ HPLC (Phenomenex LUNA C18(2), 5 μ m, Shimadzu, 10×250 mm; eluent, 15% CH₃CN/0.1% TFA; flow rate, 2 ml/min; UV detection at 205 nm) to give compound **1a** (1.8 mg) and 1b (1.3 mg).

Compound 1a. $[\alpha]_{27}^{27} = -20^{\circ}$ (*c* 0.3, MeOH) (with no TFA), ¹H NMR (CD₃OD with TFA) δ 0.95 (3H, brd, *J*=3.1 Hz, H-21), 1.08 (3H, brd, *J*=3.1 Hz, H-10), 1.25 (1H, m, H-19a), 1.30 (1H, m, H-2a), 1.35 (3H, brd, *J*=5.8 Hz, H-22), 1.64 (1H, m, H-15a), 1.67 (1H, m, H-17a), 1.71 (2H, m, H-13a, 14a), 1.72 (1H, m, H-2b), 1.73 (1H, m, H-6a), 1.76 (1H, m, H-7), 1.78 (1H, m, H-17b), 1.95 (3H, m, H-13b, 14b, 19b), 1.98 (1H, m, H-18), 2.06 (1H, m, H-6b), 2.12 (1H, m, H-3a), 2.15 (1H, m, H-15b), 2.66 (1H, dd, *J*=14.3, 8.0 Hz, H-11a), 2.80 (1H, brd, *J*=14.1 Hz, H-3b), 2.93 (3H, s, H-23), 3.04 (1H, brd, *J*=13.0 Hz, H-11b), 3.20 (1H, brt, *J*=11.5 Hz, H-1a), 3.55 (1H, brd, *J*=9.9 Hz, H-1b), 3.75 (1H, m, H-5), 3.86 (1H, m, H-20), 3.89 (1H, m, H-8), 3.90 (1H, m, H-12), 3.95 (1H, brd, *J*=11.8 Hz, H-16).

Compound 1b. $[\alpha]_{D}^{26} = +0.4^{\circ}$ (c 0.4, MeOH) (with TFA), ¹H NMR (CD₃OD with TFA) δ 0.95 (3H, d, J=6.2 Hz, H-21), 1.06 (3H, d, J=6.4 Hz, H-10), 1.32 (1H, m, H-6a), 1.35 (1H, m, H-19a), 1.39 (3H, d, J=6.1 Hz, H-22), 1.51 (1H, brd, J=14.6 Hz, H-13a), 1.63 (1H, m, H-15a), 1.65 (1H, m, H-17a), 1.72 (1H, m, H-7), 1.77 (3H, m, H-14, 17b), 1.85 (1H, m, H-13b), 1.94 (1H, m, H-19b), 1.97 (1H, m, H-2a), 1.98 (1H, m, H-18), 1.99 (1H, m, H-8a), 2.04 (1H, m, H-8b), 2.07 (1H, m, H-2b), 2.17 (1H, m, H-15b), 2.35 (1H, m, H-6b), 2.52 (1H, brd, J=12.6 Hz, H-11a), 2.91 (3H, s, H-23), 3.17 (1H, t, J=12.2 Hz, H-11b), 3.37 (1H, m, H-1a), 3.55 (1H, t, J=12.0 Hz, H-1b), 3.88 (1H, m, H-16), 3.89 (1H, m, H-12), 3.92 (1H, m, H-20), 4.17 (1H, m, H-5), 4.90 (1H, m, H-3).

1.3.6. *p*-Bromobenzoate of 8α -hydroxysenepodine A (1a). To a solution of 1a (0.45 mg) in CH₂Cl₂ (200 µl) was added *p*-bromobenzoyl chloride (5.0 mg), *N*,*N*dimethylamino pyridine (0.1 mg), and Et₃N (3 µl). The mixture was allowed to stand at 20°C for 12 h. The residue was dissolved in CHCl₃ and washed with Na₂CO₃ aq. After evaporation of solvent, the residue was applied to LH-20 column (CHCl₃/MeOH 1:1) to give a compound 1c (0.1 mg). When 1b was treated according to the same procedure as described above, the dehydration derivative 1d (0.2 mg) was produced.

Compound 1c. FABMS m/z 543 (M+H)⁺; HRFABMS m/z 543.2579 (M+H; calcd for C₃₀H₄₄N₂O₂Br, 543.2587). UV (MeOH) λ_{max} 245 (ε 10000), 212 (ε 10000): CD (MeOH) $\Delta \varepsilon_{244}$ -9.2 and $\Delta \varepsilon_{203}$ -35.5; ¹H NMR (CD₃OD) δ 0.92 (3H, d, *J*=6.6 Hz), 1.14 (3H, brd, *J*=5.1 Hz), 1.29 (3H, brs), 2.40 (3H, s), 2.66 (1H, dd, *J*=10.3, 6.4 Hz), 2.71 (1H, brt, *J*=12.5 Hz), 2.91 (1H, brd, *J*=14.6 Hz), 2.97 (1H, brd, *J*=11.7 Hz), 5.55 (1H, brs), 7.70 (2H, d, *J*=8.6 Hz), 8.04 (2H, d, *J*=8.6 Hz).

Compound 1d. ¹H NMR (CD₃OD) δ 0.93 (3H, d, J=6.4 Hz), 1.05 (3H, d, J=7.0 Hz), 1.18 (3H, brs), 2.14 (1H, brd, J=19.0 Hz), 2.42 (3H, s), 2.65 (2H, m), 2.80 (1H, t, J=12.4 Hz), 2.91 (1H, dd, J=11.4, 5.9 Hz), 5.54 (1H, brs), 5.91 (1H, brs).

1.3.7. Reduction of senepodine B (2). To a solution of 2 (0.1 mg) in MeOH (100 μ l) was added sodium borohydride (0.2 mg). The mixture was allowed to stand for 5 min. The residue was dissolved in CHCl₃ and washed with Na₂CO₃ aq. After evaporation of solvent, the residue was applied to LH-20 column (CHCl₃/MeOH 1:1) to give a compound (0.1 mg), which was identical with senepodine A (1).

1.3.8. Methylation of senepodine C (3) with MeI. A solution of 3 (0.1 mg), MeI (0.1 ml), and K_2CO_3 (0.1 mg) in acetone (0.1 ml) was heated under reflux for 6 h, and then concentrated under reduced pressure. The residue was dissolved in CHCl₃ and washed with sat. NaCl aq., and then dried over anhydrous Na₂SO₄. Removal of the solvent afforded the compound (0.1 mg), which was identical with senepodine A (1).

1.3.9. Reduction of senepodine D (4). To a solution of 4 (0.5 mg) in THF (125 μ l) was added lithium aluminum hydride (3 mg). The mixture was allowed to stand at 70°C for 4 h. After addition of EtOAc, H₃PO₄ buffer, and Na₂CO₃ aq. the mixture was extracted with CHCl₃. After evaporation of solvent, the residue was applied to C₁₈ HPLC (Mightysil C-18, 5 μ m, Kanto Chemical Co., 4.6×250 mm; eluent, 25% CH₃CN/0.1% TFA; flow rate, 1 ml/min; UV detection at 205 nm) to give a compound (0.3 mg), which was identical with senepodine A (1).

1.3.10. Hydrolysis of senepodine D (4). Compound 4 (5.0 mg) in 1N HCl (100 μ l) was allowed to stand at 100°C for 6 h. After evaporation of solvent, the residue was applied to C₁₈ (Phenomenex LUNA C18(2), 5 μ m, Shimadzu, 10×250 mm; eluent, 25% CH₃CN/0.1% TFA; flow rate, 2 ml/min; UV detection at 210 nm) to give a compound (4.0 mg), which was identical with senepodine C (3).

1.3.11. Acetylation of senepodine C (3). To a solution of **3** (4.0 mg) in pyridine (1 ml) was added acetic anhydride (0.25 ml). The mixture was allowed to stand for 12 h. After addition of H₂O, the mixture was extracted with CHCl₃. After evaporation of solvent, the residue was applied to C₁₈ HPLC (Mightysil C-18, 5 μ m, Kanto Chemical Co., 4.6×250 mm; eluent, 35% CH₃CN/0.1% TFA; flow rate, 1 ml/min; UV detection at 205 nm) to give a compound (4.0 mg), which was identical with senepodine E (**5**).

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References

- For reviews of the Lycopodium alkaloids, see: (a) Ayer, W. A.; Trifonov, L. S. *The Alkaloids*; Cordell, G. A., Brossi, A., Eds.; Academic: New York, 1994; Vol. 45, p 233. (b) Ayer, W. A. *Nat. Prod. Rep.* **1991**, *8*, 455. (c) MacLean, D. B. *The Alkaloids*; Brossi, A., Ed.; Academic: New York, 1985; Vol. 26, p 241. (d) MacLean, D. B. *The Alkaloids*; Manske, R. H. F., Ed.; Academic: New York, 1973; Vol. 14, p 348. (e) MacLean, D. B. *The Alkaloids*; Manske, R. H. F., Ed.; Academic: New York, 1968; Vol. 10, p 305.
- (a) Hemscheidt, T.; Spenser, I. D. J. Am. Chem. Soc. 1996, 118, 1799–1800. (b) Hemscheidt, T.; Spenser, I. D. J. Am. Chem. Soc. 1993, 115, 3020–3021.
- Liu, J. S.; Zhu, Y. L.; Yu, C. M.; Zhou, Y. Z.; Han, Y. Y.; Wu, F. W.; Qi, B. F. *Can. J. Chem.* **1986**, *64*, 837–839.
- (a) Cassayre, J.; Gagosz, F.; Zard, S. Z. Angew. Chem. Int. Ed. 2002, 41, 1783–1785. (b) Sha, C.-K.; Lee, F.-K.; Chang, C.-J. J. Am. Chem. Soc. 1999, 121, 9875–9876. (c) Williams, J. P.; St. Laurent, D. R.; Friedrich, D.; Pinard, E.; Roden, B. A.; Paquette, L. A. J. Am. Chem. Soc. 1994, 116, 4689–4696.

(d) Hirst, G. C.; Johnson, T. O.; Overman, L. E. J. Am. Chem. Soc. **1993**, *115*, 2992–2993, and references cited therein.

- Morita, H.; Arisaka, M.; Yoshida, N.; Kobayashi, J. J. Org. Chem. 2000, 65, 6241–6245.
- Kobayashi, J.; Hirasawa, Y.; Yoshida, N.; Morita, H. *Tetrahedron Lett.* 2000, *41*, 9069–9073.
- Kobayashi, J.; Hirasawa, Y.; Yoshida, N.; Morita, H. J. Org. Chem. 2001, 66, 5901–5904.
- 8. Hirasawa, Y.; Morita, H.; Kobayashi, J. *Tetrahedron* **2002**, *58*, 5483–5488.
- Morita, H.; Hirasawa, Y.; Yoshida, N.; Kobayashi, J. *Tetrahedron Lett.* 2001, 42, 4199–4201.
- 10. Morita, H.; Kobayashi, J. J. Org. Chem. 2002, 67, 5378-5381.
- 11. Ayer, W. A.; Iverach, G. G. Can. J. Chem. 1964, 42, 2514–2522.
- (a) Ayer, W. A.; Browne, L. M.; Nakahara, Y.; Tori, M. *Can.* J. Chem. **1979**, 57, 1105–1107. (b) Tori, M.; Shimoji, T.; Takaoka, S.; Nakashima, K.; Sono, M.; Ayer, W. A. *Tetrahedron Lett.* **1999**, 40, 323–324. (c) Tori, M.; Shimoji, T.; Shimura, E.; Takaoka, S.; Nakashima, K.; Sono, M.; Ayer, W. A. *Phytochemistry* **2000**, 53, 503–509.
- Harada, N.; Iwabuchi, J.; Yokota, Y.; Uda, H.; Nakanishi, K. J. Am. Chem. Soc. 1981, 103, 5590–5591.
- Conformational search and molecular mechanics calculations were conducted by Macromodel program: Mohamadi, F.; Richards, N. G. J.; Guida, W. C.; Liskamp, R.; Lipton, M.; Caufield, C.; Chang, G.; Hendrickson, T.; Still, W. C. *J. Comput. Chem.* **1990**, *11*, 440–467.
- 15. Halgren, T. J. Am. Chem. Soc. 1990, 112, 4710-4723.